

## THIN-FILM SOLAR CELLS BASED ON ANTIMONY CHALCOGENIDES THROUGH CHEMICAL BATH DEPOSITION

SIMPOSIO INTERDISCIPLINARIO

EN MATERIALES



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## **INTRODUCTION**

The search of new semiconductor materials for electrical energy generation constituted by non-toxic and abundant elements using inexpensive processes has been investigated in recent years. Antimony chalcogenides like antimony sulfide (Sb<sub>2</sub>S<sub>3</sub>) and antimony selenide (Sb<sub>2</sub>Se<sub>3</sub>) have been positioned as novel and attractive materials for the absorber layer in solar cells due to their interesting optoelectronic properties. Sb<sub>2</sub>S<sub>3</sub> has high optical bandgap ( $E_g$ =1.5-1.8 eV) but low electrical photoconductivity (10<sup>-8</sup>-10<sup>-6</sup>  $\Omega$ <sup>-1</sup> cm<sup>-1</sup>). On the contrary, Sb<sub>2</sub>Se<sub>3</sub> possesses  $E_g$  of 1.3-1.7 eV and higher photoconductivity than Sb<sub>2</sub>S<sub>3</sub> (>10<sup>-6</sup>  $\Omega$ <sup>-1</sup> cm<sup>-1</sup>). For this reason, the evaluation of Sb<sub>2</sub>S<sub>3</sub> in solar cells gives high open-circuit voltage ( $V_{oc}$ ) with low short-circuit current densities ( $J_{sc}$ ), while Sb<sub>2</sub>Se<sub>3</sub> produces low  $V_{oc}$  and high  $J_{sc}$  values. Therefore, the implementation of a solid solution constituted by antimony sulfide-selenide, Sb<sub>2</sub>(Se<sub>x</sub>S<sub>1-x</sub>)<sub>3</sub>, will optimize the chemical composition of the absorber to obtain both high  $V_{oc}$  and  $J_{sc}$  values in the resulting solar cells.

